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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

### **5 The Periodic Law**

modern chemistry chapter 3, Atomic math, Atomic Math. 52 terms. Atomic Structure & Isotopes. THIS SET IS OFTEN IN FOLDERS WITH... 6 terms. Chemistry Balancing Equations. 6 terms. Holt McDougal Modern Chemistry Chapter 10 Section 2 Vocabulary. 55 terms. Modern Chemistry Chapter 8. 8 terms. Holt McDougal Modern Chemistry Chapter 10 Section 3 ...

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Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT

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ANSWER Answer the following questions in the space provided. 1. \_\_\_\_ In metals, the valence electrons are considered to be (a) attached to particular positive ions.

### **Modern Chemistry Chapter 6 Section 4 Review Answers**

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