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Al $1s^2 2s^2 2p^6 3s^2 3p^1$ Si $1s^2 2s^2 2p^6 3s^2 3p^2$ P $1s^2 2s^2 2p^6 3s^2 3p^3$ 26
24 S $1s^2 2s^2 2p^6 3s^2 3p^4$ Cl $1s^2 2s^2 2p^6 3s^2 3p^5$ Ar $1s^2 2s^2 2p^6 3s^2 3p^6$ 1
1 4 2 59. 4 62. 2 65. 2 68. 2 The mixture contains at least three
parts. One solid remained on the filter paper. The other two

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Publisher Prentice Hall ISBN 978-0-13328-114-9

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within the same group of the period table behave similarly
because they have the same number of what? (octet rule,
valence electrons, electron dot structures, cations, group)
Valence electrons.

7.1 Ions Flashcards | Quizlet

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